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**Thermodynamic Study of Solarized Supercritical Carbon Dioxide Cycle**

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**ABSTRACT**

*The supercritical carbon dioxide cycle run on the utilization of carbon dioxide uses emissions from fossil fuel-fired power plants. In this study, the supercritical carbon dioxide cycle is powered by solar energy using heliostat fields. The operating range of supercritical carbon dioxide cycle permits for the integration of low-temperature cycles for effective utilization of the heat remaining unutilized in it. This paper considers the supercritical carbon dioxide cycle for its energy analysis. It aims for the optimal utilization of heat available with the exhaust from the sCO<sub>2</sub> cycle. Results obtained based on parametric variation have been presented and analyzed here. At the turbine inlet pressure of 290 bar, the turbine inlet temperature of 700oC in sCO<sub>2</sub> cycle and cycle efficiency is obtained as 16.56 % and specific work output is 174.6 KJ/kg of sCO<sub>2</sub>.*

**Keywords:** Super critical CO<sub>2</sub> cycle; Turbine inlet temperature; Specific work output; Turbine inlet pressure.

**1.0 Introduction**

Globally—the adverse impact of emissions demand for the power generation with minimum impact on the surroundings.—The capture and use of carbon dioxide for power generation through the supercritical carbon dioxide (sCO<sub>2</sub>) cycle is one of the attractive options. Supercritical carbon dioxide cycles operate on the Brayton cycle with carbon dioxide as a working fluid. Because of the constraint imposed by supercritical carbon dioxide, its exhaust carries a high amount of energy along with and can be harnessed for augmenting power output.

This section details some of the work done in this field. Turchi [1] pointed out the sCO<sub>2</sub> recompression Brayton cycle could offer the potential of equivalent or higher efficiency compared with supercritical or superheated steam cycles at temperature relevant for CSP applications. Seidel [2] compared efficiencies of several different sCO<sub>2</sub> Brayton cycle layouts (including simple regeneration cycle, recompression cycle, precompression cycle, and split expansion cycle) as the alternative power blocks in CSP systems. It was found that the recompression cycle has the best thermal efficiency over a wide range of pressure ratios. Turchi et al. [3] also examined efficiencies of different sCO<sub>2</sub> Brayton cycle layouts (including

simple regeneration cycle, recompression cycle, partial cooling cycle, and intercooling cycle) with or without reheating for CSP applications. Neises et al. [4] performed a comparison of several different sCO<sub>2</sub> Brayton cycle layouts with an emphasis on CSP applications. They compared the cycle efficiencies and the ability to integrate the thermal storage between the simple regeneration cycle, the recompression cycle, and the partial cooling cycle. Their results showed that the partial cooling cycle could offer higher efficiency than the recompression cycle until large quantities of conductance were modeled. The literature review shows that there is a need for further analysis of the supercritical carbon dioxide cycle for better energy utilization in it. Thermodynamic analysis of the solar-powered sCO<sub>2</sub> cycle operating with the steam Rankine cycle and organic Rankine cycle is pertinent for The partial cooling cycle could also offer a larger temperature regeneration cycle, recompression cycle, partial cooling cycle, and intercooling cycle) integrated with SPT systems. power cycle, and performed aexergy analysis of several sCO<sub>2</sub> Brayton cycles (including simple difference across the heat exchangers, allowing for more cost efficient thermal storage. Considering the transient nature of the solar resource, Iverson et al. [5] investigated the transient response

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of sCO<sub>2</sub> Brayton cycles to fluctuating thermal input, respectively. Selection of a suitable cycle layout is an important topic for the application of sCO<sub>2</sub> Brayton cycles in the SPT system. The literature review showed that the performances of different sCO<sub>2</sub> Brayton cycles have been compared for solar power tower applications in the above studies. However, these studies were confined to investigate the sCO<sub>2</sub> Brayton cycle separately. Few studies considered the unique characteristics of the solar receiver which have great effects on the performances of the whole system. It was normally assumed as a heat source that has similar heat outputs to those of the solar receiver. Recently, Padilla RV et al. [6] conducted a thermodynamic comparison of different sCO<sub>2</sub> Brayton cycles (including simple Brayton cycle, simple regeneration cycle, precompression cycle, split expansion cycle, and the recompression cycle) integrated with SPT systems. In their study, both the solar receiver and the heliostat field were taken into account. They concluded that the recompression studying the effect of different parameters on the performance of this combination.

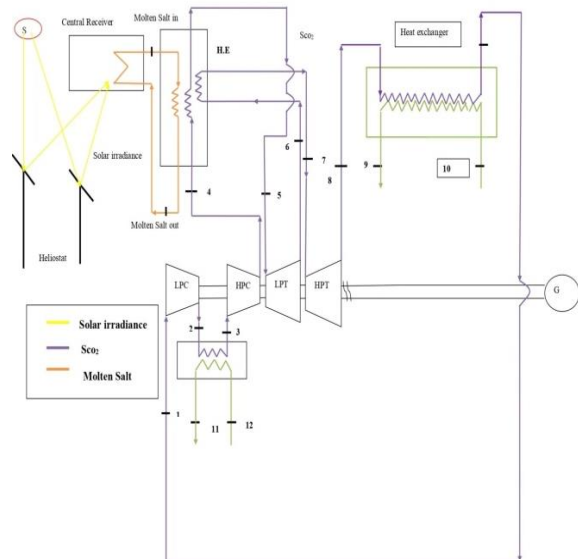
## 2.0 System description of sCO<sub>2</sub> Power System

Figure 1 is the schematic diagram of heliostat field-based solar-powered intercooler, reheat type sCO<sub>2</sub> cycle. The heliostat reflects solar radiation to the central reservoir carrying molten salt. In the present analysis, a mixture of NaNO<sub>3</sub> (59.66%) and KNO<sub>3</sub> (40.44%) is used as the salt. The boiling temperature of the salt is 1400°C and the melting temperature is 310°C to 335°C, while the heliostat field offers temperature up to 1500°C [5]. Carbon dioxide enters the LPC at state 1 for being compressed to state 2. For perfect intercooling, a heat exchanger is used between states 2 and 3 such that the temperature of state 3 is similar to that at state 1.

The compressed carbon dioxide then enters to HPC at state 3 for getting compressed up to the state 4. The compressed supercritical carbon dioxide then enters a molten salt heat exchanger, where it is heated using molten salt heat to state 5. The high pressure and high-temperature supercritical carbon dioxide enters at state 5 and then expanded in the HPT to state 6. Then the sCO<sub>2</sub> is reheated to state 7 in the molten salt heat exchanger and expanded in LPT to state 8. The expanded supercritical carbon dioxide

is sent to the heat exchanger to utilization of remaining heat at state after heat exchanger the sCO<sub>2</sub> exit at state 1 and then enter to LPC.

**Figure 1: Solarized Combined sCO<sub>2</sub> cycle with Intercooling and Reheating Arrangement in the sCO<sub>2</sub> Cycle**



## 3.0 Thermodynamic modeling

Thermodynamic modeling [14,13,7 and 15] of combined cycle components has been carried out based on the first law of thermodynamics. The following assumptions are considered for thermodynamic modeling.

1. All processes considered here are at a steady-state.
2. Direct Normal Irradiation is considered constant with a value of 1000 W/m<sup>2</sup>.
3. No variation is considered in chemical energy of the material, kinetic energy and potential energy
4. No pressure loss is considered.
5. Specific heat of supercritical CO<sub>2</sub> is taken from REFPROP 9.0
6. Inefficiency in compressor and turbine is considered through polytropic efficiency

### 3.1 Heliostat field and central receiver

Energy balance for heliostat field is given by-

$$Q_{\max} = Q_{\text{rec.}} + Q_{\text{lost}} \quad \dots(1)$$

$$\text{Where } Q_{\max} = I \cdot A_{\text{he}} \quad \dots(2)$$

$Q_{\max}$  is the total heat available in a given area ( $A_{\text{he}}$ ) on incident solar radiation ( $I$ ).

Due to some environmental factors (Irreversibility, conduction, convection, and radiation) a part of the heat is lost and remaining goes to receiver (solar isolation). After considering energy losses, the energy efficiency of heliostat ( $\eta_h$ ) is given by-

$$\eta_h = Q_{rec} / Q_{max} \quad \dots(3)$$

Energy analysis for the receiver is given by-

$$Q_{rec} = Q_{rec, abs} + Q_{rec, loss} \quad \dots(4)$$

Where  $Q_{rec, abs}$  is the absorbed energy by receiver and  $Q_{rec, loss}$  is energy lost in emission, reflection, convection and conduction.

The efficiency of receiver is given as,

$$\eta_{rec} = Q_{rec, abs} / Q_{rec} \quad \dots(5)$$

Specific heat of the molten salt in kJ/kg.K is given as,

$$C_{p,sa} = 0.172T + 1443 [7] \quad \dots(6)$$

The rate of molten salt absorbed heat can be expressed as:

$$Q_{rec, abs} = m_{sa} \cdot C_{p,sa} \cdot (T_{sa,out} - T_{sa,in}) \quad \dots(7)$$

$$Q_{rec, loss} = Q_{loss,cond} + Q_{loss,conv} + Q_{loss,ems} + Q_{loss,ref} \quad \dots(8)$$

The receiver efficiency is given as-

$$\eta_{rec} = Q_{rec, abs} / Q_{rec} \quad \dots(9)$$

$$Q_{total} = Q_{max} \cdot \eta_h \cdot \eta_r \quad \dots(10)$$

### 3.2 sCO<sub>2</sub> cycle

The sCO<sub>2</sub> cycle is the Brayton cycle with supercritical carbon dioxide as its working fluid.

### 3.3 Compressor

Considering the polytropic efficiency of the compressor

$$\eta_{poly-c} = (h_2 - h_1) / (h_2' - h_1) \quad \dots(11)$$

Work required for LPC is,

$$w_{c1} = h_2' - h_1 \quad \dots(12)$$

For perfect intercooling,

$$T_3 = T_1 \quad \dots(13)$$

Intercooling heat is,

$$Q_{int} = m_s \cdot (h_2' - h_3) = m_{int} \cdot (h_{17} - h_{18}) \quad \dots(14)$$

Between states 3 and 4,

$$\eta_{poly-c} = (h_4 - h_3) / (h_4' - h_3) \quad \dots(15)$$

Work required for HPC is,

$$w_{c2} = h_4' - h_3 \quad \dots(16)$$

Total work required ( KJ/kg) for the compressor is

$$w_{cnet} = w_{c1} + w_{c2} \quad \dots(17)$$

### 3.4 Turbine

Considering polytropic efficiency,

$$\eta_{poly-Turb} = (h_5 - h_6') / (h_5 - h_6) \quad \dots(18)$$

Work obtained from HPT is

$$w_{T1} = h_5 - h_6' \quad \dots(19)$$

For perfect reheating,

$$T_5 = T_7 \quad \dots(20)$$

Between states 7 and 8,

$$\eta_{poly-Turb} = (h_7 - h_8') / (h_7 - h_8) \quad \dots(21)$$

Work obtained from LPT is

$$w_{T2} = h_7 - h_8' \quad \dots(22)$$

Total work available from turbines is

$$w_{Tnet} = w_{T1} + w_{T2} \quad \dots(23)$$

Mass flow rate in sCO<sub>2</sub> cycle

$$Q_{total} = m_s (h_5 - h_4') + m_s (h_7 - h_6') \quad \dots(24)$$

work output of the topping cycle is given by-

$$w_{s,net} = w_{Tnet} - w_{cnet} \quad \dots(25)$$

supercritical carbon dioxide cycle's energy efficiency is expressed by-

$$\eta_s = (m_s \cdot w_{s,net}) / Q_{total} \quad \dots(26)$$

A computer program is written in C language for performance assessment with various input parameters using above equations.

## 4.0 Results and Discussion

Results are obtained from the thermodynamic modeling and computer simulation of the solarized intercooled-reheat sCO<sub>2</sub> cycle for carbon-free power for the input parameters given in Table 1 and the property values from REFPROP and e-Thermo.

**Table 1: Input Parameters [3,4,11,12,13 and14]**

Parameters	Symbol, Unit	Value
Solar irradiation	I, W/m <sup>2</sup>	1000
Heliostat field area	A <sub>hf</sub> , m <sup>2</sup>	10000
Heliostat efficiency	$\eta_h$ , %	75
Receiver efficiency	$\eta_{rec}$ , %	75
Polytropic efficiency of compressor	$\eta_{poly-c}$ , %	89
Polytropic efficiency of turbine	$\eta_{poly-turb}$ , %	93
Inlet temperature of compressor	T <sub>1</sub> , °C	32
Inlet pressure of compressor	P <sub>1</sub> , bar	75
Inlet pressure of turbine	P <sub>5</sub> , bar	250, 260, 270, 280, 290
Inlet temperature of turbine	T <sub>5</sub> , °C	600, 625, 650, 675, 700
Cycle pressure ratio	-	3.33, 3.46, 3.6, 3.7, 3.86

**Figure 2(a-e): Variation of Specific Work Output with Different TIT of sCO<sub>2</sub> Cycle**

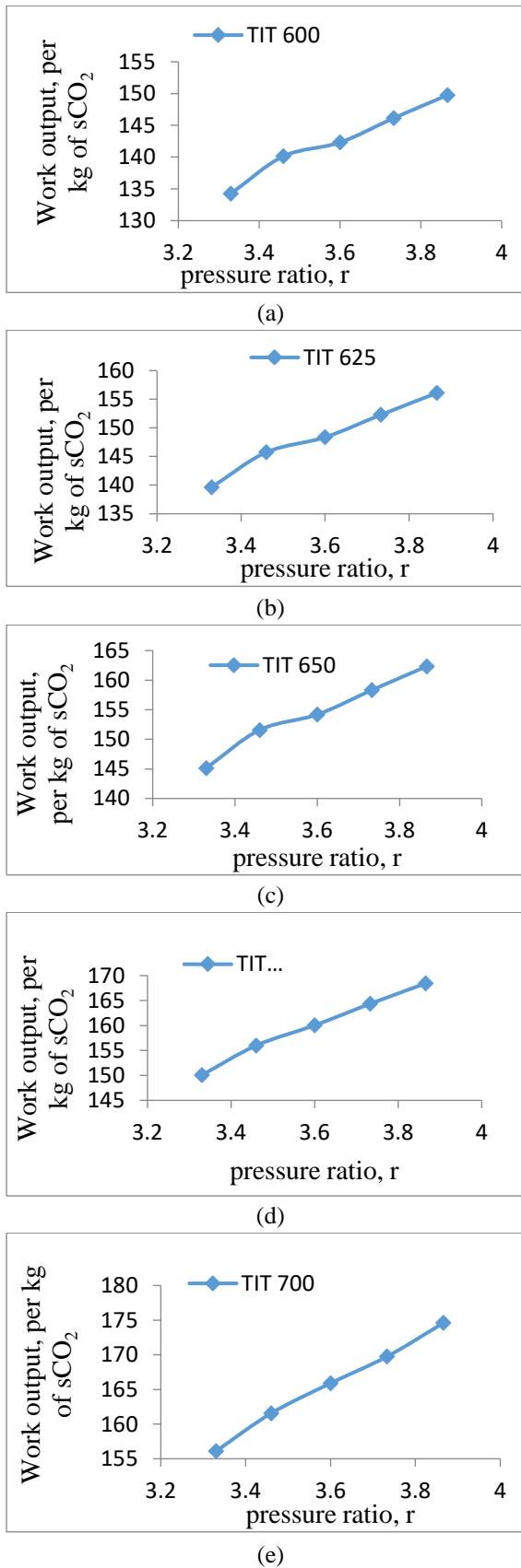
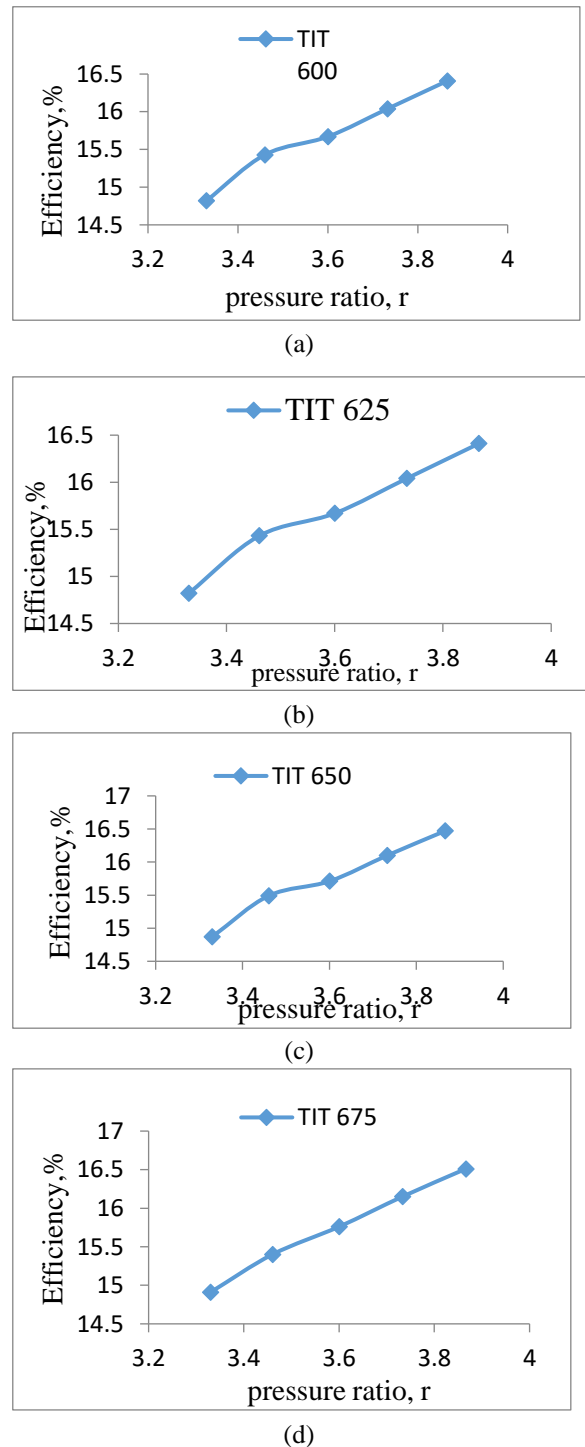
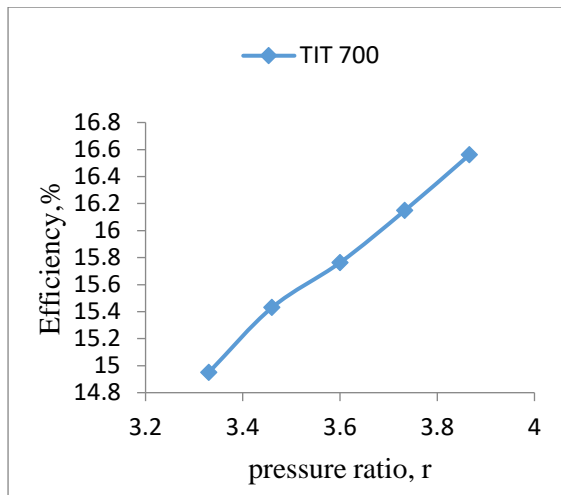


Figure 2 a-e show the variation of different work out put with cycle pressure ratio of sCO<sub>2</sub> cycle varying as 3.33,3.46,3.6,3.73,3.86. The work out put of sCO<sub>2</sub> cycle, is increases with an increase in cycle pressure ratio. The maximum work out put for sCO<sub>2</sub> cycle is obtained as 174.6 kJat 260 bat and 700 ° C .

**Figure 3 (a-e): Variation of Thermal Efficiency with TIT of the sCO<sub>2</sub> cycle for Maximum Pressures of sCO<sub>2</sub> Cycle**





(e)

Figure 2 a-e show the variation of different efficiencies with cycle pressure ratio of sCO<sub>2</sub> cycle varying as 3.33,3.46,3.6,3.73,3.86. The efficiency of sCO<sub>2</sub> cycle, is increases with an increase in cycle pressure ratio. The maximum efficiency for sCO<sub>2</sub> cycle 16.56% at 290bar pressure, 700 °C temperature at the inlet to sCO<sub>2</sub> turbine.

**5.0 Conclusions**

The following conclusions have been drawn from the thermodynamic study of solarized supercritical carbon dioxide cycle.

1. There is no emission of carbon as the supercritical carbon dioxide is the working fluid in the sCO<sub>2</sub> cycle, so it is an eco-friendly cycle.
2. There is a limitation of the sCO<sub>2</sub> cycle as the limiting pressure and temperature values are 73.773bar and 31.13°C for being in the supercritical stage.
3. In the sCO<sub>2</sub> the maximum overall thermal efficiency is 16.56% and the specific work output is 174.6 kJ per kg of sCO<sub>2</sub>.

**6.0 Acknowledgement**

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**Nomenclature**

1, 2, 3..12	Cycle states as shown in the schematic diagram
T <sub>sa,in</sub>	Molten salt inlet temperature to the central receiver
I	Solar radiation
A <sub>he</sub>	Heliostat Area, m <sup>2</sup>
av	Availability
T <sub>sa,out</sub>	Molten salt outlet temperature from central receiver
Cond.	Condenser
CPR	Cycle Pressure Ratio
CSP	Concentrated solar power
E	Energy
G	Generator
h	Enthalpy
HPC	High-Pressure compressor
LPC	Low-Pressure compressor
HPT	High-Pressure turbine
LPT	Low-Pressure turbine
LPST	Low-pressure steam turbine
TIT	Turbine inlet temperature
m	Mass
P	Pressure
p	Pump
Q	Heat
rec	Receiver
s	Entropy
sCO <sub>2</sub>	Supercritical carbon-di-oxide
T	Temperature
Turb	Turbine
MSHE	Molten salt heat exchanger
w	Work per unit mass
W	Work
η	Efficiency
<b>Subscripts</b>	
air	Air
abs	Heat absorbed by molten salt
amb.	Ambient
c	Compressor
T	Turbine
C <sub>p</sub>	Specific heat
cond.	Condenser
h <sub>e</sub>	Heliostat
isen	Isentropic
rec.	Receiver
int.	Intercooling
loss	Heat loss
m <sub>s</sub>	A mass flow rate of supercritical carbon dioxide
poly_c	Polytropic for compressor
poly_turb	Polytropic for turbine
st.	Steam
sa	Molten salt
s.	Supercritical carbon dioxide
int.	Inter cooling

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